

Review

N-Methyl antitumour agents

A distinct class of anticancer drugs?

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Summary. This article reviews the structure-activity characteristics, mode of action, pharmacokinetics and clinical utility of a group of chemically dissimilar antitumour agents which have as a common structural feature the *N*-methyl moiety. The importance of this feature is shown by the fact that molecules without a substituent on the nitrogen or compounds with *N*-alkyl groups other than methyl are usually inactive in experimental systems. This observation is supported by structure-activity studies with *N*-alkyl derivatives of *s*-triazines, triazenes, formamides, hydrazines and nitrosoureas. Representatives of these structural types which have found clinical application are, respectively, hexamethylmelamine, dacarbazine, *N*-methylformamide, procarbazine and streptozotocin. Mode of action studies have shown that dacarbazine, procarbazine and streptozotocin can give rise to species capable of methylating nucleic acid. This may be the lesion which produces antitumour activity. The mechanism of action of *N*-methylmelamines and *N*-methylformamide remains unclear. There is good evidence that, with the exception of *N*-methylnitrosoureas, host metabolism is prerequisite for activity with these agents.

Although not pronounced, the clinical activity of *N*-methyl antitumour agents is useful, particularly as activity is not associated with severe haematological toxicity. Furthermore, responses may be observed in patients resistant to bifunctional alkylating agents. It is concluded that the drugs reviewed herein show a degree of coincidence in terms of their biological properties which may warrant a common classification. The term *N*-methyl antitumour agent is proposed.

Introduction

The subdivision of anticancer drugs into various classes of agents forms the basis for the ordered study of cancer chemotherapy. Of the various classes, alkylating agents comprise the first used and probably the most widely studied type of compound. In early work it was recognised that the biological activity of alkylating agents was markedly increased by the presence of two or more alkylating functions in the molecule [131]. A range of polyfunctional compounds have subsequently found widespread clinical use,

namely, nitrogen mustards, aziridines, methane sulphonates and epoxides. More recently, it has been shown that the apparently monofunctional chloroethylnitrosoureas are indirectly bifunctional, being capable of secondary alkylation following the initial reaction of the chloroethyl carbonium ion [88]. However, there remains a group of compounds, sometimes referred to as non-classical alkylating agents [176], where there is no evidence for bifunctionality either in their structure or in their mechanism of action.

Despite the disparate range of chemical structures which can be placed in this grouping, i.e. *s*-triazines, triazenes, formamides, hydrazines and methylnitrosoureas, a feature common to the more potent representatives of these structural types is the *N*-methyl moiety. Thus, clinically used examples of these groups are, respectively, hexamethylmelamine, dacarbazine, *N*-methylformamide, procarbazine and streptozotocin. The purpose of this paper is to review the structure-activity relationships, mode of action, pharmacokinetics and clinical properties of the antitumour agents which possess the *N*-methyl moiety. On the basis of these data conclusions are drawn as to the similarities, or otherwise, between compounds, particularly with regard to the importance of the *N*-methyl group for activity and its relevance to the mechanism of action.

Structure-activity studies, metabolism, pharmacokinetics and mechanism of action

Methylmelamines

Hexamethylmelamine (HMM; Fig. 1, I) is an *s*-triazine derivative which has shown limited activity against a number of experimental rodent tumours [66, 134]. The antitumour activity of HMM is inherently associated with the presence of *N*-methyl moieties; substitution with other alkyl groups, such as ethyl, or with hydrogen leads to compounds which are devoid of antitumour activity [94, 134]. Furthermore, activity is related not only to the absolute presence of *N*-methyl groups, but also to the extent to which the compounds undergo oxidative *N*-demethylation [134]. Thus, within the homologous series, mono-, di- and trimethylmelamine (Fig. 1, V) are poorly demethylated *in vitro* and devoid of antitumour activity *in vivo* [134]. In contrast, tetramethylmelamine (Fig. 1, IV) and in particular pentamethylmelamine (PMM; Fig. 1, III) are well demethylated *in vitro* and active *in vivo* [134]. This latter observation, coupled with the greater aqueous solubility of PMM, led to

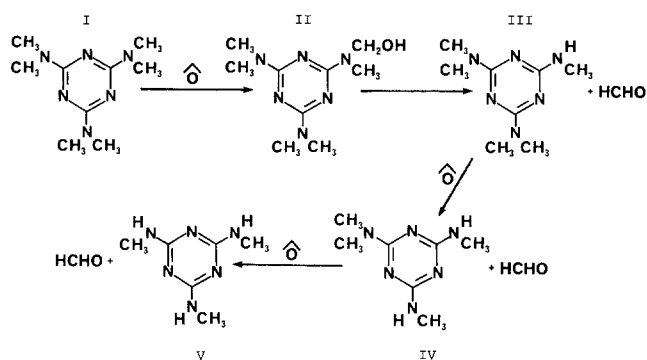


Fig. 1. Methylmelamine metabolism: I, hexamethylmelamine; II, *N*-hydroxymethylpentamethylmelamine; III, pentamethylmelamine; IV, tetramethylmelamine; V, trimethylmelamine

the selection of PMM as an alternative to HMM suitable for intravenous use (see below).

HMM and PMM are rapidly and extensively metabolised in mice, rats and humans. Thus, the majority of the drug administered is excreted in the urine in the form of demethylated metabolites, i.e. *N*², *N*⁴, *N*⁶-trimethylmelamine (Fig. 1, V), *N*², *N*⁴-dimethylmelamine, monomethylmelamine and melamine [182, 183], with very little parent drug being present [4, 25, 135]. HMM is metabolised in vitro, in the presence of various hepatic and intestinal preparations, via oxidative *N*-demethylation which yields PMM after the decomposition of an unstable carbinolamine intermediate (Fig. 1) [18, 41, 59, 134]. The intermediate produced during the *N*-demethylation of HMM to PMM, *N*-hydroxymethyl PMM (Fig. 1, II), has been identified as a major in vitro metabolite [59]. Although the formation of this same metabolite in vivo could not be demonstrated, Rutty et al. [135, 137] have provided indirect evidence for the formation of formaldehyde-precursor metabolites from HMM and PMM in vivo. More recently, Borm et al. [18] have shown that glutathione may be involved in the metabolism of HMM. If, as suggested by these authors, *N*-hydroxymethyl PMM is subject to glutathione conjugation, this may explain the failure to detect *N*-hydroxymethyl PMM in vivo.

In patients, D'Incalci et al. [36] have reported considerable variability in plasma concentration and half-life for HMM following oral administration. This does not appear to be due to poor absorption of the drug, as very little drug-derived material is excreted in the faeces [183], the majority being eliminated in the urine [4, 183]. Studies in rabbits [4] and rats [87] have confirmed that there is extensive first-pass metabolism, and hence probably activation.

Despite the fact that HMM has been used clinically for some 20 years, the mechanism by which it exerts its antitumour action remains unknown. Although *N*-methylmelamines do not have direct alkylating activity, it has been suggested that intermediate metabolites, i.e. *N*-hydroxymethylmelamines or formaldehyde, may possess such activity [94, 134, 182]. Rutty and Connors [134] have shown that whilst HMM requires metabolism in order to exert a significant cytotoxic effect in vitro, *N*-hydroxymethylmelamines are highly cytotoxic per se. *N*-Hydroxymethylmelamines are also active antitumour agents in vivo [33, 66, 71, 94, 134]. Rutty and Connors concluded, therefore, that HMM and PMM exert their antitumour effect via the formation of *N*-hydroxymethyl metabolites [134]. Further-

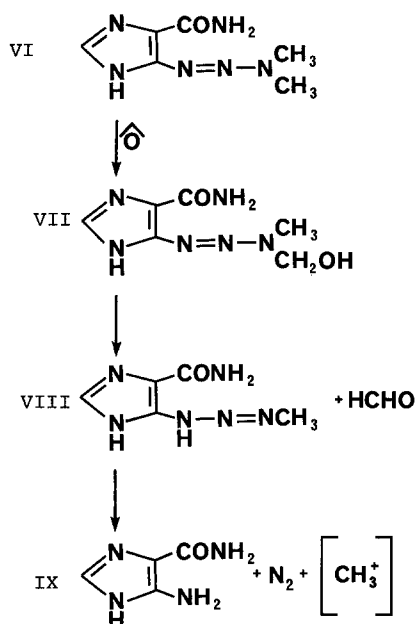


Fig. 2. Dacarbazine metabolism: VI, dacarbazine; VII, 5-(3-hydroxymethyl-3-methyltriazen-1-yl)imidazole-4-carboxamide; VIII, 5-(3-methyltriazen-2-yl)imidazole-4-carboxamide; IX, 5-aminoimidazole-4-carboxamide

more, in tumour cell lines which are sensitive to these agents in vivo, the observed cytotoxicity appears to be due to the *N*-hydroxymethylmelamine itself and not to the formaldehyde it releases on chemical decomposition [133]. Although there is evidence that HMM is itself weakly cytotoxic in vitro [37, 133], the concentrations and duration of exposure necessary to produce this effect are greatly in excess of those seen in vivo in man [36, 38, 39] and in experimental animals [4, 21, 87].

HMM and PMM have been shown to inhibit both DNA and RNA synthesis in vitro [37, 136] and in vivo [111]; however, *N*-hydroxymethylmelamines are more potent and inhibit nucleic acid synthesis irreversibly [136]. Studies with ring-¹⁴C HMM and PMM in vitro and in vivo suggest that following activation, HMM and PMM can bind to DNA [5, 6, 57, 111]. However, only limited DNA-protein cross-linking occurs either with activated HMM and PMM or with *N*-hydroxymethyl PMM in L1210 cells [132]. Furthermore, in the sensitive PC6 cell line no significant DNA cross-linking could be demonstrated [114]. Thus it seems unlikely that DNA cross-linking plays a major role in the molecular mechanism of action of the *N*-methylmelamines.

Dacarbazine

After 10 years of clinical experience with and experimental work on dacarbazine (DTIC; Fig. 2, VI), its mode of antineoplastic activity remains an enigma. It was synthesised in 1961 as a prodrug of 5-diazoimidazole-4-carboxamide [151], an agent which had been shown to possess antitumour activity [149, 150]. Most studies on the biological properties and structure-antitumour activity relationships of 1-aryl-3,3-dialkyltriazenes have been conducted with the 1-phenyl-substituted analogues of dacarbazine [7, 32, 60], whose chemical properties and spectrum of antineoplastic activity against rodent malignancies resemble those

of dacarbazine. In one of the first reports on the antineoplastic activity of 1-aryl-3,3-dialkyltriazenes, against the murine sarcoma 180, the structural requirement of *N*³-methyl substitution in the triazene molecule was demonstrated [29]. More recently, in an investigation of the activity of 35 different 1-aryl-3,3-dialkyltriazenes against the mouse TLX/5 lymphoma, Connors et al. [32] deduced that only those arylalkyltriazenes that can be metabolised in vivo to an arylmonomethyltriazene possess antitumour properties. However, the diethyl analogue of dacarbazine and several related compounds with alkyl moieties other than methyl have shown marginal activity against the L1210 leukaemia [70].

In addition to their activity against primary tumours, it has been shown by Giraldi, Sava and co-workers that certain 1-aryl-3,3-dimethyltriazenes can inhibit the metastatic spread of tumours in mice [63, 64, 140–142], and dacarbazine also appears to possess this property [30].

There is now convincing evidence that 1-aryl-3,3-dimethyltriazenes require metabolic activation by the host for antitumour activity. Most of the results corroborating this contention have been obtained in studies on phenyl dimethyltriazenes rather than on dacarbazine [7, 32, 60]. These triazenes undergo oxidative *N*-demethylation catalysed by cytochrome P450 to form 1-aryl-3-monomethyltriazenes, direct methylating agents. For dacarbazine this metabolic route is outlined in Fig. 2. Aminoimidazole carboxamide (AIC; Fig. 2, IX) is the major urinary metabolite of dacarbazine in man [19, 75, 76, 158, 159] and is formed from dacarbazine in mouse liver microsomes [73] and human and animal tumour tissue [58, 107]. It is difficult to envisage from the mechanistic point of view how AIC could be formed by a metabolic pathway other than one implicating 5-(3-methyl-triazene-2-yl)imidazole-4-carboxamide (MTIC; Fig. 3, VIII). Recently, Kolar et al. [90] identified 5-(3-hydroxymethyl-3-methyltriazene-1-yl)imidazole-4-carboxamide (HMTIC; Fig. 2, VII) as a urinary metabolite of dacarbazine in rats. As this compound appeared to be more stable than MTIC in polar solvents, the authors suggested that HMTIC may act as a transport form of MTIC, the postulated ultimate antineoplastic species derived from dacarbazine. An alternative transport form of MTIC might be the hydroxymethyltriazene glucuronide of HMTIC, given the observation that the hydroxymethyltriazene glucuronide is a metabolite of 1(2,4,6-trichlorophenyl)-3,3-dimethyltriazene in rats [89].

MTIC decomposes spontaneously in aqueous media to form AIC, with the concomitant alkylation of nucleophiles [118], e.g. DNA [106]. Recently, however, evidence has been presented which casts doubt on the contention that an indiscriminantly reactive species such as MTIC can be responsible for the selective antitumour activity of dacarbazine [60, 70, 72].

The most extensive study of dacarbazine pharmacokinetics in man reported to date has shown that AIC and the parent compound can be readily detected in both plasma and urine [19], confirming earlier studies which showed that renal excretion is a major route of elimination [76, 100, 157–159]. Studies currently underway have shown that both HMTIC and MTIC are dacarbazine metabolites in mice, rats and man, although differences apparently exist between species in terms of the formation and persistence of these metabolites [138]. Finally, the observation that 7-methylguanine excretion is enhanced in patients

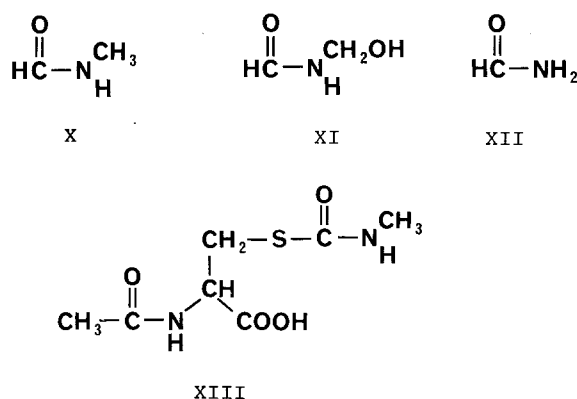


Fig. 3. *N*-methylformamide and its metabolites: *X*, *N*-methylformamide; *XI*, *N*-hydroxymethylformamide; *XII*, formamide; *XIII*, *S*-(*N*-methylcarbamoyl)*N*-acetylcysteine

and rats treated with dacarbazine confirms that this agent can give rise to methylating species in vivo [160].

N-Methylformamide

The antitumour activity of *N*-methylformamide (NMF; Fig. 3, X) against murine tumours was first described in the early 1950s. In studies of the antitumour efficacy of a series of formamide derivatives against the S180 sarcoma [28] and Ehrlich ascites tumour [55], NMF was found to be the most potent inhibitor of tumour growth. This result has been confirmed recently using two other murine tumour models, although formamide (Fig. 3, XII), did display marginal antitumour activity in these systems [61]. Interestingly, *N,N*-dimethylformamide, like *N*-ethylformamide, is only slightly active or inactive in most rodent antitumour tests [95], although both DMF and NMF similarly inhibited the growth of two human colon tumours grown in immune-deprived mice [45].

The mechanism by which NMF exerts its antitumour activity is under discussion. The work of Clarke et al. [28] suggests that NMF is not a folic acid antagonist. Further studies indicated that NMF may interfere with nucleic acid synthesis [28, 48, 113, 139, 161, 178], although it appeared to stimulate the incorporation of formate into nucleic acids in the liver [9], an organ to which it is toxic [117]. A rodent tumour which had acquired resistance to aryl-dimethyltriazenes and procarbazine in vivo was not resistant to NMF [61] and hence NMF apparently has a different mechanism of action. NMF and related polar solvents can induce terminal differentiation to a more mature phenotype in certain human leukaemia cell lines, and the suggestion has been tendered that this is related to their antineoplastic activity in mice [164].

Whether NMF requires bioactivation in order to exert an antineoplastic effect is difficult to establish, as it is not metabolised to an appreciable extent in vitro by liver preparations [61]. However, the fact that *N*-ethylformamide, although devoid of any in vivo antitumour activity, exhibited toxicity in vitro comparable to that of NMF [61] suggests that metabolism may be a determinant of the selective in vivo antitumour activity of NMF. Evidence that NMF may be bioactivated to reactive metabolites in vivo has emerged from studies of its effect on hepatic non-protein thiols. Of a series of formamide derivatives, only NMF caused a reduction in hepatic non-protein thiols, an

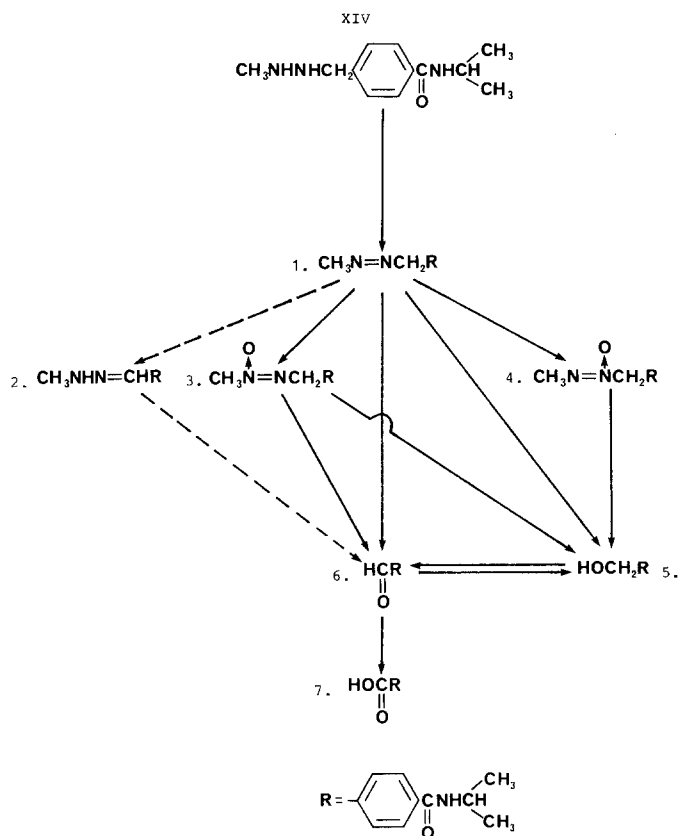


Fig. 4. Metabolism (—) and chemical decomposition (-----) of procarbazine (XIV): 1. azoprocarbazine; 2. *N*-isopropyl-*p*-formylbenzamide methylhydrazone; 3. methylazoxyprocarbazine; 4. benzylazoxyprocarbazine; 5. *N*-isopropyl-*p*-hydroxymethylbenzamide; 6. *N*-isopropyl-*p*-formylbenzamide; 7. *N*-isopropyl-terephthalamic acid

effect which could be partially abolished by pretreatment of animals with SKF-525A [61]. This finding implicates the formation of a reactive — presumably electrophilic — metabolite in vivo.

Metabolism studies in mice have shown that NMF is biotransformed to CO_2 , which is exhaled in the breath, and to methylamine, *N*-hydroxymethylformamide (HMF; Fig. 3, XI), formamide (Fig. 3, XII) and a mercapturic acid conjugate, all of which are excreted in the urine [85]. The latter metabolite has subsequently been identified as *S*-(*N*-methylcarbamoyl)*N*-acetylcysteine (Fig. 3, XIII) and was also found in the urine of patients treated with NMF [86]. Although HMF is unlikely to contribute to the antitumour activity of NMF [34], the as yet unknown precursor of the mercapturic acid may be a cytotoxic species.

After the administration of NMF or $^{14}\text{CH}_3\text{-NMF}$ to mice the plasma concentration vs time profiles did not fit a linear pharmacokinetic model, and low levels of radioactivity were measurable in the plasma for up to 8 days [20, 22].

Procarbazine

Procarbazine (Fig. 4, XIV) was originally synthesised in the laboratories of Hoffman-La Roche following the observation that 1-methyl-2-benzylhydrazine displayed antitumour activity in experimental systems [13, 184]. Although the data has never been presented, it has been stat-

ed that only 1-methylhydrazines possess antitumour activity, ethyl and unsubstituted hydrazines being inactive [13, 91]. Early workers showed that procarbazine is not cytotoxic to cultured cells [69], and this in turn stimulated metabolism studies in an attempt to identify an active metabolite. As these studies have progressed, it has become apparent that the metabolism and chemical decomposition of procarbazine are particularly complicated, with some 15 different metabolites or intermediates being implicated to date [35, 153].

The chemical decomposition of procarbazine involves oxidation to azoprocarbazine and subsequent isomerisation of the double bond to form the methylhydrazone which hydrolyses to yield *N*-isopropyl-*p*-formylbenzamide (Fig. 4) [67, 175]. Although it was originally proposed that this pathway was also the major route of procarbazine metabolism [130], more recent data has shown that the methylhydrazone is not a major product of procarbazine metabolism and that azoprocarbazine is relatively stable under physiological conditions [8, 93, 110, 175]. More importantly, the in vitro oxidation of procarbazine to azoprocarbazine can be catalysed by various liver preparations [8, 93, 110, 175], a process which is probably dependent on cytochrome P450 [47]. The subsequent metabolism of azoprocarbazine can apparently follow four routes, namely, direct conversion to *N*-isopropyl-*p*-formylbenzamide or *N*-isopropyl-*p*-hydroxymethylbenzamide [35] and *N*-oxidation to produce two azoxyprocarbazine isomers (Fig. 4) [175, 179]. In vitro the formation of the methylazoxy isomer is quantitatively the most important reaction [35, 179], and in rat and human plasma this compound is the major primary azoprocarbazine metabolite [153]. Both of the azoxy metabolites can be converted in vitro by hepatic preparations to *N*-isopropyl-*p*-hydroxymethylbenzamide or, in the case of the methylazoxy isomer, the corresponding aldehyde (Fig. 4) [35, 175]. All of the above reactions are probably mediated by cytochrome P450 [35, 179] and, for the *N*-oxidation of azoprocarbazine, the isoenzymes responsible have been identified [128]. Finally, *N*-isopropyl-*p*-formylbenzamide and *N*-isopropyl-*p*-hydroxymethylbenzamide are oxidised to *N*-isopropyl-terephthalamic acid (Fig. 4) the major urinary metabolite [122, 130]. Urinary excretion in man, dogs and rats is extensive, with up to 75% of the dose excreted in the urine within 24 h [143].

Recently, it has been shown that azoprocarbazine metabolism in vitro is associated with methyl-radical-derived methane generation [129]. The methyl radical is responsible, in part, for the methylation of microsomal proteins during in vitro azoprocarbazine metabolism. In addition, electron spin resonance studies have directly demonstrated the existence of carbon-centred free radicals during procarbazine metabolism [156]. These reactive species may therefore be involved in mediating procarbazine antitumour activity; however, the exact nature of the metabolite responsible is currently unknown. In addition to procarbazine, azoprocarbazine and a mixture of the two azoxy isomers possess in vivo antitumour activity [14, 152], an observation which is consistent with the hypothesis that antitumour activity is related to the presence of an intact *N*-methyl moiety.

A number of studies using methyl-radiolabelled procarbazine have confirmed that extensive metabolism of the methyl group occurs in vivo. Thus radioactivity is expired

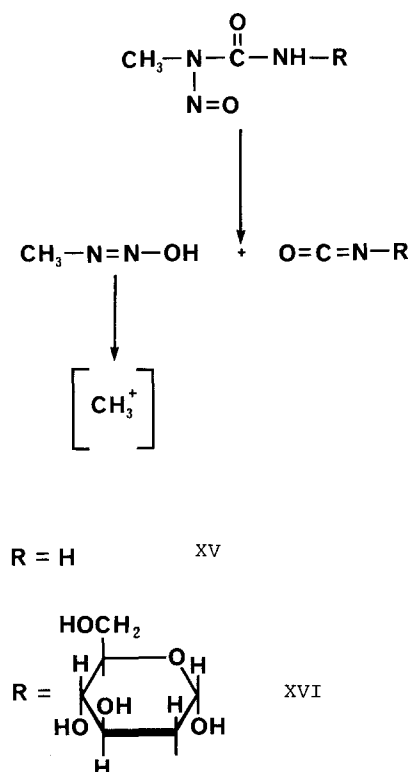


Fig. 5. The chemical decomposition of methylnitrosoureas: XV, methylnitrosourea; XVI, streptozotocin

by rats in the form of CO_2 and methane [46, 143]. In addition to elimination, there is incorporation of radioactivity into the DNA and RNA of tumour cells from mice treated with procarbazine [24, 91]. Two major routes of incorporation into nucleic acid have been identified; direct methylation of bases, involving complete transfer of the intact *N*-methyl group, and incorporation during de novo purine biosynthesis via labelled formate [91]. Although the most thorough study has involved RNA [91], DNA methylation has also been shown [24].

N-Methylnitrosoureas

Apart from the well-known *N*-chloroethylnitrosoureas (BCNU, CCNU, MeCCNU, etc.) two *N*-methylnitrosoureas have also received clinical evaluation, i.e. *N*-methylnitrosourea (MNU Fig. 5, XV) and streptozotocin (Fig. 5, XVI). Although MNU is known to western science primarily as an experimental carcinogen [78] it has been used in the USSR as an antitumour agent (see below). Streptozotocin is a fermentation product which was initially developed as an antibiotic [171]. Following the demonstration of activity in experimental systems [51], it was hoped that, because of its low bone marrow toxicity [126], streptozotocin would have greater utility than *N*-chloroethylnitrosoureas. Unfortunately, clinical trials have shown that the activity of streptozotocin is limited to certain rare tumour types (see below).

Early structure-activity studies demonstrated that nitrosourea antitumour activity resides only in the *N*-methyl and *N*-chloroethyl analogues [82]. This is in marked contrast to the carcinogenic activity of nitrosoureas, where there is an extensive literature documenting the activity of,

for example, *N*-ethyl- [77] and *N*-propylnitrosoureas [168], as well as MNU [78].

No definite evidence for the metabolic activation of either MNU or streptozotocin has been reported. However, both compounds are known to undergo spontaneous chemical decomposition to yield alkylating species [126]. The decomposition pathway for MNU and streptozotocin, shown in Fig. 5, leads to the release of a methylating species which may be the ultimate cytotoxic species.

Studies on the mechanism of action of MNU have focussed mainly on the DNA lesion responsible for carcinogenesis [96]. A number of methylated bases have been isolated from nucleic acid extracts of organs from animals treated with MNU. Quantitatively, the most important methylated base is *N*⁷-methylguanine, although evidence from a number of studies suggests that *O*⁶-methylation of guanine is the lesion which leads to carcinogenesis via anomalous base pairing [78, 96].

More recent studies on the mechanism of action of methylating agents have focussed on the importance of DNA repair enzymes as determinants of cell sensitivity. For example, the enzyme responsible for the repair of *O*⁶-methylguanine (*O*⁶-methylguanine-DNA methyltransferase) is absent in cells which are particularly sensitive to the cytotoxic effects of methylating agents [154]. However, it does not necessarily follow that *O*⁶-methylguanine is the sole cytotoxic lesion, since the extent of *O*⁶-guanine methylation does not always correlate with cell survival [119].

N-Methylnitrosourea-induced changes in poly (ADP-ribose) polymerase, a second DNA repair enzyme, may be an additional determinant of activity in certain human tumours. Treatment of cultured cells with MNU or streptozotocin stimulates poly (ADP-ribose) polymerase, resulting in an increased utilisation, and thereby depletion, of the substrate nicotinamide adenine dinucleotide (NAD) [162]. NAD depletion is thought to be the reason why *N*-methylnitrosoureas induce diabetes in experimental animals, as pretreatment with nicotinamide prevents the onset of diabetes [144]. In experimental systems nicotinamide pretreatment does not impair the antileukaemic activity of streptozotocin, and hence, in this situation, the mechanism of action is probably alkylation alone [144]. Clinically, however, the activity of streptozotocin against islet cell carcinomas may be due in part to the diabetogenic properties of the molecule (see below).

Clinical utility

Methylmelamines

Due to its poor aqueous solubility HMM is given as an oral preparation, usually as a daily dose over several weeks. Whereas it can cause leukopaenia and thrombocytopenia, these effects are usually mild, and the common limiting toxicity is patient intolerance due to cumulative nausea and vomiting. The other important side effects of HMM are neurological [101, 181] and may be ameliorated by pyridoxine [52].

HMM has a wide spectrum of antitumour activity, responses being seen in breast, ovarian, lung and cervical carcinomas [11, 97], lymphomas [124] and sarcomas [17]. Unfortunately, its activity is poor in patients already treated with standard drugs. Thus, of 104 patients with breast cancer resistant to conventional therapy only two responded [52, 98]. In ovarian cancer the response rate is

30%–40% in patients not previously treated [97, 177], but in disease resistant to nitrogen mustard alkylating agents it is modest: 15% [12], 16% [123], 27% [16] and 29% [81], giving an overall response rate of 19%. However, Vogl et al. [173] found a response rate of 54% to an HMM-cisplatin combination in melphalan-resistant disease. Ovarian carcinoma is the only disease where HMM is used with any frequency.

PMM, being water-soluble can be given intravenously, thereby circumventing variation in bioavailability (see above). Furthermore, it was hoped that the drug-induced nausea would be less than that seen with HMM. In fact it is considerably worse, and is the limiting toxicity. It has been difficult to produce any myelotoxicity with this drug. In a number of phase I studies, nearly 200 patients have been treated, most with unassessable disease. Only eight minor responses and one partial response have been seen [3, 26, 65, 80, 115, 170]. In a comparative trial of PMM and HMM for the treatment of bilharzial bladder cancer, the two agents were similar in terms of toxicity, response and survival [56].

Dacarbazine

Dacarbazine has unquestioned activity in Hodgkin's disease, sarcomas and melanoma. However, its most notable side effect, nausea and vomiting, has made it an unpopular drug. Dacarbazine has been used most extensively in malignant melanoma, reviewed by Comis [31]. As a single agent it consistently produces remissions in the range of 16%–31%. Its superiority over other agents in this disease is not established, however [1, 2].

In a single-agent study [54] of pretreated patients with Hodgkin's disease, 10 of 18 patients responded to dacarbazine. Hence it is now used frequently as second-line therapy in this disease in the ABVD regimen [15]. In non-Hodgkin's lymphomas, remissions are fewer and shorter [54].

A number of studies have shown the activity of dacarbazine against a variety of sarcomas [53, 102, 172, 174]. Gottlieb et al. [68] reviewed the sequential M. D. Anderson and Southwest Oncology Group studies. The addition of dacarbazine to adriamycin apparently increased the response rate for sarcomas from 31% to 42%. The survival of the patients treated in the combination studies was significantly longer than that of those in the single-agent adriamycin study.

Nearly all patients receiving dacarbazine experience nausea and vomiting [102]. When given on a daily $\times 5$ schedule it is frequently observed that these effects are worst on the first day. Leukopaenia is variable and usually less severe than thrombocytopenia.

Dacarbazine is always given intravenously, although it has been given by mouth [100, 157] in daily doses sufficient to produce myelosuppression. Variability in absorption [100, 157] caused the intravenous route to be preferred.

N-Methylformamide

A number of clinical trials with *N*-methylformamide have been described [50, 105, 121, 167, 180]. Mild nausea is seen at higher doses, which commonly also cause elevations of liver enzyme and bilirubin levels. Hepatotoxicity or general malaise are dose-limiting; little activity has been seen.

Procarbazine

The clinical use of procarbazine was reviewed in 1974 by Spivack [163]. No new role has been found for the drug since then.

Single-agent activity in Hodgkin's disease was found early on [104] in both previously untreated disease and disease resistant to other agents [40, 74, 169]. The inclusion of procarbazine in four-drug combinations with a nitrogen mustard, vinca alkaloid and prednisone has proved highly successful in the management of advanced Hodgkin's disease, producing complete remission rates of over 80% for previously untreated patients [43, 84, 120]. Interestingly, the combination of cyclophosphamide, vincristine and prednisone, omitting procarbazine, gave an equivalent figure of only 31% [103].

Similarly, single-agent activity against non-Hodgkin's lymphomas [99, 169] has led to procarbazine's inclusion in a successful four-drug regimen (cyclophosphamide, vincristine, prednisone and procarbazine) [44]. Good, possibly better results have been obtained, however, with several drug combinations which exclude procarbazine [10, 147].

The use of procarbazine in lung cancer remains controversial. An early report [148] of a 67% response rate in a small number of previously untreated patients with small cell histology has led to its inclusion in a number of drug combinations for this disease. Its efficacy has recently been questioned, however, by Gersel Pedersen et al. [62], who observed no responses in a phase II single-agent study in 24 patients with disease resistant to other agents.

Although procarbazine can produce shrinkage of non-small cell lung carcinomas in about 18% of cases, the remission is brief in duration [99] and confers no survival benefit [125].

Procarbazine is usually given as a daily oral dose. Mild nausea and vomiting are frequently seen initially, but tend to subside. Myelotoxicity is dose-limiting. Peripheral neuropathy and effects on the central nervous system are occasionally seen. Flushing after alcohol is a fairly common interaction. As procarbazine is a weak monoamine oxidase inhibitor [42], the usual exaggerated pressor responses to sympathomimetics and tyramine-containing foods occur.

The intravenous route of administration has not been directly compared with the oral route. Those who have used low repeated doses i.v. have found comparable myelosuppression and equivalent efficacy [104, 112, 155]. Following intermittent high intravenous doses, central nervous system effects are dose-limiting, with prolonged somnolence and disorientation followed by restlessness and emotional lability in the recovery period. Nausea and vomiting may be severe, and peripheral neuropathy is also more pronounced [27].

N-Methylnitrosoureas

Streptozotocin has been used with some success in islet cell tumours of the pancreas [23, 116]. Whether its efficacy is related to the diabetogenic effect seen in some animal species [92, 144] is impossible to say, since there are no reports of other nitrosoureas used against this tumour. The improvement in the response rate seen when 5-fluorouracil is added to streptozotocin [108] suggests that islet cell tumours may be relatively chemosensitive. Activity against carcinoid tumours is less impressive [109, 145, 166]. Re-

sponses of short duration have been seen in Hodgkin's disease resistant to conventional chemotherapy. In one instance tumour shrinkage was seen in a patient resistant to BCNU [146].

The limiting toxicity of streptozotocin is renal, very little myelosuppression being seen at the doses used.

MNU has been used clinically in the USSR [49]. Good responses were reported in undifferentiated lung cancer and previously treated Hodgkin's disease. Almost all patients experienced severe nausea and vomiting, often accompanied by diarrhoea, within 40 min of receiving the drug. Myelotoxicity was also seen.

Conclusions and future developments

From the extensive literature concerning the clinical and experimental properties of the antitumour agents reviewed herein it is possible to draw several conclusions. Structure-activity studies have shown that in the majority of *in vivo* rodent antitumour screens, only compounds with an *N*-methyl moiety display significant activity. Obvious exceptions to this generalisation are the *N*-chloroethylnitrosoureas, the *N*-chloroethyltriazenes and the aziridiny melamine triethylenemelamine. These agents are, however, distinct from their *N*-methyl analogues by virtue of their potential for bifunctional alkylation. Although inadequate absorption, metabolism or distribution may be invoked to explain the inactivity of compounds with *N*-alkyl groups other than methyl, e.g. hexaethylmelamine [21], it seems unlikely that this is sufficient as a general explanation. For example, *N*-ethylnitrosourea has potent *in vivo* biological activity in terms of carcinogenicity [77] yet is ineffective against an MNU-sensitive tumour [82]. Similarly, *N,N*-diethyltriazenes are carcinogenic [127] yet induce little or no tumour inhibition [29, 32, 70]. Thus, on balance, it appears that the *N*-methyl group is an important determinant of antitumour activity.

In marked contrast to the stringent requirement for activity at the *N*-alkyl position are the wide range of nitrogen-containing structures which can bear the *N*-methyl group, i.e. triazines, triazenes, formamides, hydrazines and nitrosoureas. Although it might be suggested that the requirement for an *N*-methyl moiety in these various classes of molecules implies a common mechanism of action, caution should be exercised. Whilst it is true that *N*-methyltriazenes, *N*-methylnitrosoureas and procarbazine can give rise to methylating species, there is no evidence that methylmelamines or NMF can do so. If the former compounds do in fact act by methylation of nucleic acid, then for methylmelamines and NMF the structural requirement for a methyl group is apparently coincidental. Therefore, on the basis of the available data, a common mechanism of action for all of the *N*-methyl drugs is not implicated, a view which is supported by the observed differences in their activities in murine antitumour screens [66]. However, there is still no convincing evidence that any of these compounds act via a mechanism involving bifunctional alkylation.

With the exception of nitrosoureas, all of the agents reviewed herein probably require metabolic activation, although in the case of NMF the evidence is somewhat circumstantial. These observations argue strongly for the application of clinical pharmacokinetic studies to demonstrate that adequate metabolic activation is achieved in man. It has been suggested on the basis of such studies

that, in the case of methylmelamines, the direct administration of the *N*-hydroxymethyl metabolic intermediate may result in greater antitumour activity in man [137]. Preliminary clinical results with one such compound, *N*²,*N*⁴,*N*⁶-*tris* (hydroxymethyl)-*N*²,*N*⁴,*N*⁶-trimethyl melamine, indicate that *N*-hydroxymethylmelamines are active antitumour agents in man [83]. Similarly, a novel class of monoalkyltriazene prodrugs has recently been described which again do not require metabolic activation in order to display activity [165]. This approach is currently the most exciting area of drug development in the field of *N*-methyl antitumour agents.

Clinically, it must be admitted that *N*-methyl drugs have not had a major impact on cancer chemotherapy. However, their redeeming features are that they do not produce severe haematological toxicity and that in some instances they are not cross-resistant with nitrogen mustard alkylating agents. These properties have made them valuable components of many combination schedules, particularly in second-line therapy.

With regard to disease-specific activity in man, although there is some overlap between compounds, e.g. in lymphomas, adequate comparative single-agent studies have not been reported. Hence it is difficult to comment on the similarities, or otherwise, between the various agents. In terms of toxicity, although nausea and vomiting is the side effect most frequently observed, the most worrying long-term effect is that of carcinogenicity. For example, it is accepted that dacarbazine, procarbazine and methylnitrosoureas are carcinogenic in man [78, 79].

In summary, the antitumour agents reviewed in this article show a degree of coincidence in terms of structure-activity characteristics, mode of action and metabolic activation. Notable exceptions to this generalisation are the weak activity of certain dialkyltriazenes without an *N*-methyl group, the lack of nucleic acid methylation following HMM or NMF treatment and the chemical activation of methylnitrosoureas. Despite these discrepancies, none of the compounds appear to act via a polyfunctional mechanism, and as such they clearly differ from classical bifunctional alkylating agents. Furthermore, their *in vivo* properties distinguish them from the latter class of compounds. Thus, in conclusion, the description of this group of compounds as *N*-methyl antitumour agents may form the basis of a more rational classification.

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